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out the amounts of reactants and products that take part in a chemical reaction. To begin, check that this equation is shown:  $Fe_2O_3 + 3CO \rightarrow 2Fe + 3CO_2$ . Look at the coefficients (such as the “ 3 ” in  $3CO$ ) in front of each substance in the equation.

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1. Look at the coefficients (such as the “ 3 ” in  $3\text{CO}$ ) in front of each substance in the

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Stoichiometry (Show all required parts) Use the following to answer questions 1 & 2.  $\text{NaCl} + \text{MgO} \rightarrow \text{Na}_2\text{O} + \text{MgCl}_2$ .

1. If 24 grams of sodium chloride reacts with an excess amount of magnesium oxide, how many grams of sodium oxide will be produced?

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Answers To

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Worksheet

What is the molar mass  
of  $C_3H_4$ ? 2 mol  $O_2$ :1  
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