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Solution: $p_{\text{abs}} = 140 \text{ kPa}$ $1 \text{ } 14:70 \text{ psi}$ $101:3 \text{ kPa} = 20:32 \text{ psia}$ $p_{\text{gage}} = p_{\text{abs}} - p_{\text{atm}} = (20:32 \text{ psia}) - (14:70 \text{ psia}) = 5:62 \text{ psi}$ $p_{\text{gage}} = 5:62 \text{ psig}$ c.) Situation: Pressure values need to be converted. Find: Calculate the absolute pressure (psia) corresponding to a pressure of 0.55 bar (gage). Properties: $p_{\text{atm}} = 14:70 \text{ psi}$
Solution: $p_{\text{gage}} = 0:55 \text{ bar}$ $1 \text{ } 14:70 \text{ psi}$ $1:013 \text{ bar} = 7:98 \text{ psig}$ p

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