

Photochemistry Of Organic Compounds From Concepts To Practice

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Photochemistry of Organic Compounds: From Concepts to Practice Petr Klán and Jakob Wirz. © 2009 P. Klán and J. Wirz. ISBN: 978-1-405-19088-6. Postgraduate Chemistry Series. A series designed to provide a broad understanding of selected growth areas of chemistry. at postgraduate student and research level.

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Organic photochemistry encompasses organic reactions that are induced by the action of light. The absorption of ultraviolet light by organic molecules often leads to reactions. In the earliest days, sunlight was employed, while in more modern times ultraviolet lamps are employed. Organic photochemistry has proven to be a very useful synthetic tool. Complex organic products can be obtained simply.

~~Organic photochemistry - Wikipedia~~

Indeed, one of the classic photochemical reactions of organic chemistry is the formation of 1,1,2,2-tetraphenyl-1,2-ethanediol (3, benzopinacol) by the action of light on a solution of diphenylmethanone (2, benzophenone) in isopropyl alcohol. The yield is quantitative.

~~28.3: Organic Photochemistry - Chemistry LibreTexts~~

Abstract Examines the electronics of the simple carbonyl group, the effect of structure on photoreduction of benzophenones

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and acetophenones, non-conjugated unsaturated ketones, the photochemistry of cyclic enones, and cross-conjugated cyclohexadienones.

~~Photochemistry of organic compounds: II, Carbonyl ...~~

Typical absorption range of some important classes of organic compounds: Simple alkene 190 - 200 nm . Acyclic diene 220 - 250 nm . Cyclic diene 250 - 270 nm . Styrene 270 - 300 nm . Saturated ketones 270 - 280 nm . α,β -Unsaturated ketones 310 - 330 nm . Aromatic ketones/aldehydes 280 - 300 nm

~~Photochemistry—Organic Syntheses with Light~~

□ The first law of photochemistry, the Grotthuss-Draper law, states that light must be absorbed by a compound in order for a photochemical reaction to take place. □ The second law of photochemistry, the Stark-Einstein law, states that for each photon of light absorbed by a chemical system, only one molecule is activated for subsequent reaction. This "photoequivalence law" was derived by ...

~~Photochemistry—Michigan State University~~

Photochemistry is the branch of chemistry concerned with the chemical effects of light. Generally, this term is used to describe a chemical reaction caused by absorption of ultraviolet, visible light or infrared radiation. In nature, photochemistry is of immense importance as it is the basis of photosynthesis, vision, and the formation of vitamin D with sunlight. Photochemical reactions proceed differently than temperature-driven reactions. Photochemical paths access high energy intermediates th

~~Photochemistry—Wikipedia~~

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