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CHAPTER 22, Hydrocarbon

Compounds(continued) 14. Circle the letter of each condensed structural formula for pentane. a.  $C_5H_{12}$  b.  $CH_3CH_2CH_2CH_2CH_3$  c.  $CH_3(CH_2)_3CH_3$  d.  $C-C-C-C$  15. The IUPAC system uses \_\_\_\_\_ to show the number of carbon

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SECTION 22.1 HYDROCARBONS(pages  
693 – 701)

744 Chapter 21 • Hydrocarbons Section  
221.11.1 Introduction to Hydrocarbons  
MAIN Idea Hydrocarbons are carbon-

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containing organic compounds that provide a source of energy and raw materials. Real-World Reading Link If you have ridden in a car or a bus, you have used hydrocarbons. The gasoline and diesel fuel that are used in cars, trucks, and ...

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types of hydrocarbons using four carbon atoms: a. two noncyclic alkanes b. two cycloalkanes 3.

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– He was named after the actor Sean  
Connery, who played James Bond in the  
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What are Hydrocarbons? Hydrocarbons are organic compounds containing only hydrogen and carbon atoms. For example, the gas used in gas stoves and bunsen burners can be either propane ( $C_3H_8$ ) or

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butane ( $C_4H_{10}$ ). Combustion of  
Hydrocarbons All hydrocarbons burn  
' cleanly ' to form water and carbon  
dioxide in the presence of sufficient oxygen.

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– He was named after the actor Sean  
Connery, who played James Bond in the  
60s.

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## Key Concepts 22.1 Hydrocarbons •

Because carbon has four valence electrons, carbon atoms always form four covalent bonds. • The carbon atoms in an alkane can be arranged in a straight chain or in a chain that has branches. • Molecules of hydrocarbons, such as alkanes, are nonpolar molecules. 22.2 Unsaturated Hydrocarbons



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## CHAPTER 22 Study Guide

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student is asked to write structural formulas for two different hydrocarbons having the molecular formula  $C_5H_{12}$ . She writes one formula with all five carbon atoms in a horizontal line and the other with four carbon atoms in a line, with a  $CH_3$  group extending down from the first attached to the third carbon atom.

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12.E: Organic Chemistry- Alkanes and  
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Day 1: Alkanes - You must know how to name and draw straight chain and branched alkanes. Homework: Finish the Straight Chain Alkane practice at the end of the

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notes. **GOALS:** Naming, molecular formula, structural formula, condensed formula, and carbon skeletal formula. I couldn't find another worksheet, but the workbook covers these topics.

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