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Electrochemistry Practice Test with Answers Question 1. On passing 0.5 faraday of electricity through NaCl, the amount of Cl deposited on cathode is. Related: PES Scholastic Aptitude Test Chemistry. A. 17.75 gm. B. 142 gm. C. 71 gm. D. 35.5 gm. Question 2 ...

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AP Chemistry: Electrochemistry Multiple Choice Answers 14. Questions 14-17 The spontaneous reaction that occurs when the cell in the picture operates is as follows: $2\text{Ag}^+ + \text{Cd (s)} \rightarrow 2\text{Ag (s)} + \text{Cd}^{2+}$ (A) Voltage increases. (B) Voltage decreases but remains $>$ zero.

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THE “ OFFICIAL ” CHEMISTRY 12 REDOX
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Multiple Choice Questions (Type-II) Note : In the following
questions two or more than two options may be correct. The
positive value of the standard electrode potential of Cu^{2+}/Cu indicates that _____.

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reducing agent than the H^+ / H_2 couple. (ii) this redox couple is a stronger oxidising agent than H^+ / H_2 .

Class 12 Important Questions for Chemistry – Electrochemistry

31. During electrolysis of water the volume of oxygen liberated is 2.24 dm^3 . The volume of hydrogen liberated, under same conditions will be : 2.24 dm^3 . 1.12 dm^3 . 4.48 dm^3 . 0.56 dm^3 . Answer. 32. In Down's method for the extraction of sodium, the melting point of the electrolyte is lowered by adding.

Important Questions of Electrochemistry for NEET Chemistry

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Multiple Choice Questions Answers

Answers: 1. (d) 2. (e) 3. (a) 4. (e) 5. (d) 6. (a) 7. (d) 8. (b) 9. (d)
10. (c) 11. (b) 12. (c) 13. (b) 14. (c) 15. (c) 16. (d) 17. (e) 18. (b)
19. (e) 20. (c) 21. (c) 22. (d)

Sample Questions - Chapter 21

If $E^\circ_{\text{Fe}^{2+}/\text{Fe}} = -0.441 \text{ V}$ and $E^\circ_{\text{Fe}^{2+}/\text{Fe}^{3+}} = 0.771 \text{ V}$, the standard EMF of the reaction, $\text{Fe} + 2\text{Fe}^{3+} \rightarrow 3\text{Fe}^{2+}$ will be.
(a) 1.212 V. (b) 0.111 V. (c) 0.330 V. (d) 1.653 V. Answer.

Answer: a. We hope the given Chemistry MCQs for Class 12 with Answers Chapter 3 Electrochemistry will help you.

Chemistry MCQs for Class 12 with Answers Chapter 3 ...

Multiple choice questions Try the following multiple choice questions to test your knowledge of this chapter. For each

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question there is one correct answer. The periodic table, physical constants and relative atomic masses needed for these problems are given on the inside covers of Chemistry, fourth edition by C.E. Housecroft and E.C. Constable.

Multiple choice questions - Pearson Education

Sample Questions - Chapter 21 Electrochemistry Examples of Multiple Choice Questions: 1. In an electrolytic cell the electrode at which the electrons enter the solution is called the _____; the chemical change that occurs at this electrode is called _____. (a) anode, oxidation (b) anode, reduction (c) cathode, oxidation (d) cathode, reduction

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Multiple Choice. Identify the choice that best completes the statement or answers the question. _____ 1. The half-reaction that occurs at the cathode during the electrolysis of molten sodium bromide is _____. a. $+ 2e^- 2Br^-$ Br. 2^- . b. $+ 2e^- 2Br^-$. c. $+ e^- Na^+ - Na$ d.

AP Chemistry-Electrochemistry
ElectrochemistryClass 12 Chemistry Multiple Choice

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Mcqs Of Chapter Electrochemistry

Answer. Answer: (b) 2. An increase in the conductivity equivalent of a solid electrolyte with dilution is primarily due to. (a) increased ionic mobility of ions. (b) 100 percent electrolyte ionisation with natural dilution. (c) increase in both ion numbers and ionic mobility. (d) A rise in ion counts.

MCQ on Electrochemistry - NCERT Books

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NEET Chemistry Electrochemistry Multiple Choice Questions
MCQs on Electrochemistry : 1. The aqueous solution of
which of the following compounds is the best conductor of
electric current?

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