

## Behavior Of Gases Review 2 Answers

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2. Gas particles are in random, constant, straight line motion. 3. When particles  
collide with each other (or any container), the collisions are said to be elastic. 4. The  
volume that gas particles take up is negligible. And from the distance between  
particles is relatively great.

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Student Review Packet Answer Key

Behavior Of Gases Review 2 SECTION 2 BEHAVIOR OF GASES 1. a measure of how fast the particles of an object are moving 2. when it is heated 3. Temperature of gas particles Energy of gas particles Volume of gas particles 1) 20 °C Particles have the

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smallest amount of energy. Volume is smallest. 2) 50 ° C Particles have more

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You usually cannot feel it, but air has pressure. The gases in Earth ' s atmosphere exert pressure against everything they contact. The atmosphere rises high above Earth ' s surface. It contains a huge number of individual gas particles. As a result, the pressure of the tower of air above a given spot on Earth ' s surface is substantial.

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