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Objectives 8.1.1

Describe how a
substance can

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dissolve in water by
dissociation,
dispersion, or
ionization. 8.1.2

Describe how the
physical properties of
a solution can differ
from those of its
solute and solvent.

8.1.3 Identify energy
changes that occur
during the formation
of a solution. 8.1.4

Describe factors

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Section 8.1 8.1

Formation of
Solutions

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Describe factors affecting the

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carney. Description:

Describe how a
substance can
dissolve in water by
dissociation,
dispersion, or

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ionization. Describe how the physical properties of a solution can differ from those of its solute and solvent;

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Solubility and
Concentration 8.3:
Properties of Acids
and Bases 8.4:
Strength of Acids and
Bases. Solute. a
substance whose
particles are

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dissolved in a
solution. Solvent.

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Section 8.1 8.1

Formation of

Solutions Ans. Yes.

Section 8 (1) of the

Companies Act, 2013

allows person or

association of persons

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to be registered as a Section 8 Company on the fulfillment of certain conditions and procedure as prescribed therein. The term “ person ” has not been defined in the Companies Act, 2013.

Section 8 1 Formation
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Objectives 8.1.1

Describe how a substance can dissolve in water by dissociation, dispersion, or ionization. 8.1.2

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Solutions that occur
during the formation
of a solution. 8.1.4
Describe factors
affecting the

Section 8.1 8.1
Formation of
Solutions - Physical
Science

8.1 formation of
solutions. STUDY.
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minor component in a solution, dissolved in the solvent. Solvent. able to dissolve other substances.

Dissociation. the splitting of a molecule into smaller molecules, atoms, or ions, especially by a reversible process.

Dispersion.

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solutions Flashcards |
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Section 8.1:

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8 FOCUS Objectives

8.1.1 Describe how a
substance can
dissolve in water by
dissociation,
dispersion, or
ionization. 8.1.2

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Describe how the physical properties of a solution can differ from those of its solute and solvent.

8.1.3 Identify energy changes that occur during the formation of a solution. 8.1.4

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Homogeneous
solutions are
solutions with
uniform composition

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and properties throughout the solution. For example a cup of coffee, perfume, cough syrup, a solution of salt or sugar in water etc. Heterogeneous solutions are solutions with non-uniform composition and properties throughout the solution.

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Answers

Types of Solutions -
Different Types,
Homogeneous ...

8.1 Formation of
Solutions Dissolving

~Every Solution has
two components ~a
solute and a solvent

~a solute is the
substance whos
particles are

dissolved ~a solvent is

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Answers

the substance that the
solute dissolves ~the
two components can
appear in three forms
~liquid, gas, and solid

8.1 Formation of
Solutions by Anika
Conrad

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solution to form one
substance must

dissolve in another

for this to happen the

solute and solvent

particles must attract

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during the formation
of a solution energy is
absorbed released 18
the solubility of solids
increases as

Section 8 1 Formation
Of Solutions Answer
Key

Section 8.1 8.1

Formation of
Solutions Ans. Yes.

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Section 8 (1) of the Companies Act, 2013 allows person or association of persons to be registered as a Section 8 Company on the fulfillment of certain conditions and procedure as prescribed therein.

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